

International Journal of Plant & Soil Science

Volume 35, Issue 18, Page 149-156, 2023; Article no.IJPSS.102904 ISSN: 2320-7035

Magnesium Supplementation Concerning Soil Chemical Properties and Cowpea Yield in Ultisol of Kerala, India

V. P. Soniya ^{a*} and P. S. Bhindhu ^b

^a Department of Soil Science and Agricultural Chemistry, College of Agriculture, Kerala Agricultural University, Kerala, India. ^b Soil Science and Agricultural Chemistry, ICAR-KVK, Kottayam, Kerala, India.

Authors' contributions

This work was carried out in collaboration between both authors. Both authors read and approved the final manuscript.

Article Information

DOI: 10.9734/IJPSS/2023/v35i183278

Open Peer Review History:

This journal follows the Advanced Open Peer Review policy. Identity of the Reviewers, Editor(s) and additional Reviewers, peer review comments, different versions of the manuscript, comments of the editors, etc are available here: https://www.sdiarticle5.com/review-history/102904

Original Research Article

Received: 18/05/2023 Accepted: 10/07/2023 Published: 15/07/2023

ABSTRACT

Ultisol of Kerala faces high magnesium deficiency due to poor retention of magnesium to soil exchange surfaces under high rainfall conditions. Hence magnesium nutrition management is an important aspect of fertility improvement in these soils. We conducted a pot culture study to determine the influence of varying magnesium levels on soil chemical properties and yield of cowpea in Ultisols of Kerala. We used a completely randomized design. The study was carried out in the Radiotracer Laboratory, College of Agriculture, Vellanikkara, Kerala Agricultural University, in January-April 2019. Magnesium carbonate at levels 5 to 80 mg.kg⁻¹ was provided along with recommended doses of fertilizers for cowpea maintained in pots. Rhizosphere samples were analyzed during flowering and at harvest of the crop following standard procedures. Yield and yield attributes are recorded at the time of harvest. Graded doses of magnesium introduced significant

^{*}Corresponding author: E-mail: soniya-2019-21-068@student.kau.in;

Int. J. Plant Soil Sci., vol. 35, no. 18, pp. 149-156, 2023

(P< 0.05) variations in the soil available nutrients and yield of cowpea. Soil pH and available magnesium in rhizosphere soil during flowering and post-harvest increased with the increasing levels of magnesium carbonate, and using 80 mg kg⁻¹ of magnesium recorded the highest pH, available phosphorus, potassium, and magnesium. In contrast, the gradation of magnesium could not produce significant variation in yield. The highest yield was recorded in treatment supplied with 10 mg kg⁻¹ of magnesium and was on par with those of higher levels of magnesium supplement. Graded doses of magnesium introduced variation in available nutrients but a better yield response in cowpea was obtained from magnesium at 10 mg.kg⁻¹.

Keywords: Available nutrients; magnesium; ultisol; yield.

1. INTRODUCTION

Ultisol occupy more than 50% of the total geographical area of Kerala. The soils are characterized by low pH, and low cation exchange capacity leading to low nutrient retention due to the dominance of kaolinite, oxides, and hydrous oxides of iron and aluminium [1]. Magnesium saturation constitutes 4-20% of the cation exchange capacity of the soil [2]. Even though magnesium is one of the major exchangeable cations on the exchange complex of soil, magnesium deficiency is a major constraint to crop productivity. About 90-98% of the soil Mg is incorporated in the crystal lattice minerals, thus not structure of directly available for plant uptake. Crop loss due to magnesium deficiency can be alleviated with appropriate fertilization practices. Wilkinson [3] positive reported а correlation between phosphorus and magnesium in the soil. Generally, Mg has an effect on potassium translocation in soil. Gransee and Führs [4], noted that continuous excessive potassium fertilization increases the risk of induced deficiency in magnesium soil due to magnesium interference in uptake. Barber [5] reported the existence of negative interaction between calcium and magnesium in the soil. Kene et al. [6] observed a reduced calcium uptake by plants in high magnesium-containing soil, and the plants grown under such conditions deficiency. showed calcium Studies on interactions between calcium, magnesium, and potassium showed a suppressive effect of calcium and potassium on magnesium uptake, which also depends on ionic concentration and soil properties [7]. Studying interaction interface nutrient at root-soil is an important aspect of mineral nutrition. Hence, We aimed to study the influence applied magnesium on of soil chemical properties and corresponding variation in cowpea yield.

2. MATERIALS AND METHODS

The pot culture experiment was laid out as a completely randomized design with 11 treatments and four replication in January-April 2019. Topsoil (0-15cm) representing Ultisol was collected from Water Management Research Unit (13⁰32N and 76⁰26E), Kerala Agricultural University. The soil sample was air dried, ground, sieved through a 2 mm sieve, and characterized for physicochemical properties.

Five kilograms of soil were filled in earthen pots. Treatments included were control (T_1) , recommended dose of fertilizers (RDF; T₂), RDF + magnesium (5 mg.kg⁻¹; T₃), RDF+ magnesium (10 mg.kg⁻¹; T₄), RDF+ magnesium (15 mg.kg⁻¹; T₅), RDF+ magnesium (20 mg.kg⁻¹; T₆), RDF + magnesium (30 mg.kg⁻¹; T₇), RDF+ magnesium (40 mg.kg⁻¹; T₈), RDF+ magnesium (50 mg.kg⁻¹; T₉), RDF + magnesium (60 mg.kg⁻¹; T₁₀) and RDF + magnesium (80 mg.kg⁻¹; T₁₁). Four replications of each treatments were maintained and two replications were used for destructive sampling during flowering of the crop.

The recommended dose of fertilizers for cowpea includes the application of 20 t.ha⁻¹ of (VC), 250 kg.ha⁻¹ vermicompost calcium carbonate, and 20:30:10 kg.ha⁻¹ of N, P_2O_5 , and K₂O and was modified based on initial soil test result [8]. After applying calcium carbonate, a one-week interval was given for applying organic manure. Two weeks post-application of organic manure, varying levels of magnesium carbonate were added following the treatments. Three seeds of bhagyalakshmi, a bush cowpea variety, were sown in each pot, and one healthy seedling was maintained after one week of emergence. Half a dose of nitrogen and complete doses of phosphorus and potassium were applied after thinning the population. The remaining dose of nitrogen was supplied two weeks after the first application. Foliar application of boron (0.05%) was done twice to combat boron deficiency. The nutrients were supplied through water-soluble sources. Organic manure used for the study was characterized and was found to have a pH of 7.10. The nitrogen, phosphorus, potassium, calcium, and magnesium in the VC were 1.71, 0.30, 0.61, 1.27, and 0.28% respectively (Table 2). Irrigation with de-ionized water, weed control, and plant protection measures were adopted uniformly in each pot.

Soil analysis was carried out during flowering and at the harvest of the crop. At flowering, soil samples were collected by destructive sampling of two replications. A composite sample from five pots maintained under each replication was used for analyzing various chemical properties. Soil pH, EC, K, Ca, and Mg were determined by following the procedure of Jackson [9]. The organic carbon was analyzed by the wet oxidation method of Walkley and Black [10], available N by Subbiah and Asija [11] method, P by Bray and Kurtz [12] method and Fe, Mn, Zn, and Cu was determined by following the procedure of Sims and Johnson [13]. The data were analyzed statistically using the OPSTAT software package [14] and Duncan's multiple range test was employed to test the significance of the difference between means of treatments.

3. RESULTS AND DISCUSSION

The soil was sandy clay in texture with an initial pH of 4.7. Organic carbon, available nitrogen, and potassium were medium in status, while phosphorus was high. The secondary and micronutrients except magnesium and boron were sufficient (Table 1).

3.1 Effect of Magnesium Supplementation on Soil pH

Soil pH measured during flowering and after harvest differed significantly (P<0.05) between treatments with the highest increment obtained in soil treated with 80 mg.kg⁻¹ (Table 3). Applying magnesium carbonate with a neutralizing value of 118.61% resulted in an increased soil pH. The increased soil pH observed after crop harvest compared to the flowering stage in all treatments except absolute control might be due to the slow solubility of magnesium carbonate. The solubility of dolomite/ magnesite was 87% less than kieserite three weeks after application [15].

3.2 Effect of Magnesium Supplementation on Organic Carbon Status of Soil

The organic carbon content in the soil increased in all treatments except absolute control during flowering and harvest, which can be attributed to the addition of VC (Table 3). Available nitrogen in the soil was significantly (P<0.05) higher in treatment supplied with a recommended dose of fertilizers during flowering and post-harvest, which can be attributed to the higher organic carbon status of soil in this treatment (Table 1). The organic carbon content of the soil is taken as the index of nitrogen supplying power as the C: N ratio is usually stabilized at 10:1 to 12:1 in tropical humid climates [16,17].

3.3 Effect of Magnesium Supplementation on Primary Nutrient Status of Soil

Soil analysis for available phosphorus during flowering and harvest showed a significantly (P<0.05) higher status in treatment supplied with 80 mg.kg⁻¹ magnesium, which might be due to the increase in soil pH (Table 3). Hence result was following the finding of Fageria et al. [18], who reported an increase in available phosphorus as pH increased to above 5.0, due to the release of P ions from Al and Fe oxides. Adams [19] also reported the occurrence of positive correlation and interactions between phosphorus and magnesium in soil and that Mg helps in greater solubilization of phosphorus in soil.

Available potassium was significantly (P<0.05) higher in soil received 80 mg.kg⁻¹magnesium both flowering and after harvest (Table 4). This might be due to the release of potassium from the exchange sites to maintain the equilibrium between the soil solid phase and solution phase. According to Schofield's ratio law, the ratio of cations held by the soil and the ratio in an equilibrium solution is constant [20]. Hannaway [21] studied the effect of Mg on K translocation in soil and reported that low magnesium status in soil decreases the available K.

3.4 Effect of Magnesium Supplementation on Secondary Nutrient Status in Soil

Available calcium level increased from the initial level in all treatments except absolute control due to the calcium release from calcium carbonate/ organic manure (Table 4). A further increase in available calcium in the soil was observed post-harvest, which might indicate the release of calcium from calcium carbonate and/or organic manure. Though there is conflicting information concerning the reaction time of limestone in acid soils, Jones and Mallarino [22] reported a significant influence of reagent-grade calcium carbonate in the soil after 200 days of incubation though a significant increase in pH was realized within 10 days.

The variations in available Mg content in soil at both stages of analysis corresponded to the gradation in magnesium through added sources with the highest content in treatment supplied with 80 mg.kg⁻¹ of magnesium (Table 4). An increase in available magnesium status at crop harvest when compared to the flowering stage indicates the release of magnesium from magnesium carbonate. Further, an increase in the available pool of nutrients in maintaining sufficient soil humidity and temperature was reported by Fageria [2].

Soil parameters	Value
Sand (%)	46.90
Silt (%)	11.60
Clay (%)	40.30
Texture	Sandy clay
рН	4.70
Electrical Conductivity (EC) (dS.m ⁻¹)	0.07
Organic carbon (OC) (%)	1.32
Available nitrogen (Av. N) (kg.ha ⁻¹)	476.67
Available phosphorus (Av. P) (kg.ha ⁻¹)	98.04
Available potassium (Av. K) (kg.ha ⁻¹)	240.18
Available calcium (Av. Ca) (mg.kg ⁻¹)	429.30
Available magnesium (Av. Mg) (mg.kg ⁻¹)	64.53
Available sulphur (Av. S) (mg.kg ⁻¹)	5.00
Available iron (Av. Fe) (mg.kg ⁻¹)	12.41
Available manganese (Av. Mn) (mg.kg ⁻¹)	16.26
Available zinc (Av. Zn) (mg.kg ⁻¹)	3.81
Available copper (Av. Cu) (mg.kg ⁻¹)	8.08
Available boron (Av. B) (mg.kg ⁻¹)	0.24
Effective cation exchange capacity (cmol(+) kg ⁻¹)	5.63

Table 1. Initial soil properties of the experimental site

Table 2. Characteristics of vermicompost (VC)

Parameters	Content	
pH	7.10	
EC (dS m ⁻¹)	0.81	
Nitrogen (%)	1.79	
Phosphorus (%)	0.30	
Potassium (%)	0.61	
Calcium (%)	1.97	
Magnesium (%)	0.28	
Sulfur (%)	0.25	
Iron (mg.kg ⁻¹)	1000.00	
Manganese (mg.kg ⁻¹)	290.60	
Zinc (mg.kg ⁻¹)	80.50	
Copper (mg.kg ⁻¹)	24.00	
Boron (mg.kg ⁻¹)	64.40	

Treatments		рН		Organic carbon (%)		Available Nitrogen(kg ha ⁻¹)		Available P (kg ha ⁻¹)	
		Flowering	Harvest	Flowering	Harvest	Flowering	Harvest	Flowering	Harvest
T ₁	Absolute control	4.75 ^g	4.72 ⁹	1.27 ^g	1.08 ^t	347.72 ^g	286.12 ^c	82.89 ⁹	72.23 ^h
T_2	The recommended dose of fertilizers (RDF)	4.88 ^f	4.92 ^e	2.54 ^a	2.12 ^a	534.55 ^a	501.72 ^a	122.55 ^{bc}	137.24 ^a
T_3	RDF + magnesium (5 mg kg ⁻¹)	4.94 ^d	4.96 ^{de}	2.02 ^c	1.90 ^{abc}	502.46 ^{bc}	489.21 ^a	116.89 ^d	98.28 ^e
T_4	RDF + magnesium (10 mg kg ⁻¹)	4.96 ^d	4.97 ^{de}	1.43 ^f	1.72 ^{bcd}	512.37 ^b	345.18 ^b	107.14 [†]	98.79 ^e
T_5	RDF + magnesium (15 mg kg ⁻¹)	4.88 ^{ef}	4.91 ^e	2.54 ^a	1.70 ^{cde}	503.52 ^{bc}	332.41 ^{bc}	121.50 [°]	125.1 [°]
T_6	RDF + magnesium (20 mg kg ⁻¹)	4.94 ^d	4.98 ^{cd}	1.51 [†]	1.67 ^{cde}	371.02 [†]	482.94 ^a	113.3 ^e	88.1 ^g
T_7	RDF + magnesium (30 mg kg^{-1})	5.01 ^c	5.02 ^c	1.48 ^f	1.525 ^{de}	506.38 ^b	502.94 ^a	108.35 ^f	94.48 ^f
T ₈	RDF + magnesium (40 mg kg^{-1})	5.04 ^c	5.14 ^b	1.87 ^d	1.50 ^{de}	509.52 ^b	348.45 ^b	107.69 ^f	98.29 ^e
T ₉	RDF + magnesium (50 mg kg ⁻¹)	4.92 ^{de}	4.95 ^{de}	2.23 ^b	1.47 ^e	429.98 ^e	495.48 ^a	123.83 ^b	125.20 ^c
T ₁₀	RDF + magnesium (60 mg kg ⁻¹)	5.10 ^b	5.14 ^b	1.77 ^{de}	1.95 ^{ab}	512.09 ^b	470.39 ^a	106.7 ^f	109.73 ^d
T ₁₁	RDF + magnesium (80 mg kg ⁻¹)	5.20 ^a	5.20 ^a	1.74 ^e	1.99 ^a	491.40 ^c	472.39 ^a	131.76 ^a	129.88 ^b

Table 3. Effect of Mg supplementation on soil pH, organic carbon, available N, and available P content of the soil

Treatment means with common superscripts do not differ significantly

Table 4. Effect of supplementation on available potassium, calcium, and magnesium

Treatments		Potassium (kg ha ⁻¹)		Calcium (mg kg ⁻¹)		Magnesium (mg kg ⁻¹)	
		Flowering	Harvest	Flowering	Harvest	Flowering	Harvest
T ₁	Absolute control	139.38 ^t	191.99 ^h	390.67 [°]	467.35 [†]	55.57 ¹	63.10 ⁹
T_2	The recommended dose of fertilizers (RDF)	183.29 ^b	268.24 ^{bc}	511.75 ^b	575.02 ^d	73.77 ⁹	80.90 ^e
T_3	RDF + magnesium (5 mg kg ⁻¹)	149.01 ^e	234.24 ^{fg}	502.00 ^b	580.50 ^d	67.72 ^h	70.52 ^f
T_4	RDF + magnesium (10 mg kg ⁻¹)	150.02 ^e	244.49 ^e	510.75 ^b	624.00 ^a	75.90 [†]	79.49 ^b
T_5	RDF + magnesium (15 mg kg ⁻¹)	148.06 ^e	259.56 ^{cd}	502.40 ^b	627.25 ^a	76.90 ^{fg}	93.10 ^d
T_6	RDF + magnesium (20 mg kg ⁻¹)	156.96 ^d	229.60 ⁹	504.75 ^b	598.25 [°]	77.85 [†]	82.55 ^e
T_7	$RDF + magnesium (30 mg kg^{-1})$	157.86 ^d	242.42 ^{et}	502.07 ^b	616.50 ^{ab}	81.77 ^e	89.42 ^d
T_8	RDF + magnesium (40 mg kg^{-1})	165.20 [°]	256.92 ^d	538.25 ^a	605.30 ^{bc}	93.05 ^d	100.80 ^c
T ₉	RDF + magnesium (50 mg kg ⁻¹)	179.42 ^b	250.82 ^{de}	520.00 ^{ab}	597.50 [°]	105.25 [♭]	110.40 ^b
T ₁₀	$RDF + magnesium (60 mg kg^{-1})$	166.65 [°]	276.92 ^b	499.92 ^b	614.00 ^{abc}	101.10 ^c	126.82 ^a
T ₁₁	$RDF + magnesium (80 mg kg^{-1})$	198.74 ^a	290.64 ^a	507.25 ^b	557.25 ^e	123.47 ^a	130.95 ^a

Treatment means with common superscripts do not differ significantly

Treat	ments	Plant height (cm)	Pods per plant	Length of pods (cm)	Yield (g plant⁻¹)
T ₁	Absolute control	37.90 ^t	8.25 ^t	9.60 ⁹	33.04 ^e
T_2	RDF + magnesium (5 mg kg ⁻¹)	52.80 ^c	19.5 ^{abc}	17.70 ^a	70.39 ^c
T_3	RDF + magnesium (10 mg kg ⁻¹)	61.65 ^a	19.50 ^{abc}	15.80 ^{bc}	79.33 ^a
T_4	RDF + magnesium (15 mg kg ⁻¹)	56.95 ^b	18.75 ^{abc}	15.95 ^{bc}	73.33 ^{bc}
T_5	RDF + magnesium (20 mg kg ⁻¹)	49.65 ^d	20.25 ^{ab}	15.05 ^{cd}	76.57 ^{ab}
T_6	RDF + magnesium (30 mg kg ⁻¹)	58.95 ^b	17.25 ^{bc}	17.00 ^{ab}	75.28 ^{abc}
T_7	RDF + magnesium (40 mg kg ⁻¹)	48.40 ^d	18.00 ^{bc}	15.50 [°]	73.30 ^{bc}
T ₈	RDF + magnesium (50 mg kg ⁻¹)	49.80 ^d	21.75 ^a	16.35 ^{abc}	76.12 ^{abc}
T ₉	RDF + magnesium (60 mg kg ⁻¹)	48.25 ^d	16.50 ^{cd}	17.70 ^a	74.64 ^{abc}
T ₁₀	RDF + magnesium (80 mg kg ⁻¹)	49.55 ^ª	18.75 ^{abc}	13.95 ^{de}	74.19 ^{abc}

Table 5. Effect of supplementation on yield and biometric characteristics of Cowpea

Treatment means with common superscripts do not differ significantly

3.5 Effect of Magnesium Supplementation on Yield of Cowpea

Yield and related biometric attributes were significantly (P<0.05) influenced by the varying levels of magnesium added (Table 5). Significantly (P<0.05) higher plant height was obtained in treatment supplied with 10 mg.kg⁻¹ of magnesium followed by 30 and 15 mg.kg⁻¹ of magnesium. The treatments differed significantly (P<0.05) concerning the number of pods per plant. A significantly (P<0.05) higher number of pods per plant was obtained in plants that received 50 mg.kg⁻¹ of magnesium and were on par with that of 5, 10, 15, 20, and 80 mg.kg⁻¹ of magnesium. Significantly (P<0.05) long pods were observed in plants supplied with 5, 30, and 60 ma.ka⁻¹ of magnesium. The treatments differed significantly with respect to the yield per plant. Plants treated with 10 mg.kg⁻¹ of significantly (P<0.05) magnesium recorded higher yields but were on par with that of 50, 20, 30, 60, and 80 mg.kg⁻¹ magnesium received plants. The absolute control treatment recorded the lowest yield. The lack of growth response to the higher dose of magnesium addition indicated that moderate level of magnesium ie 10 mg.kg would be sufficient to meet the magnesium requirements of cowpea [23-24].

4. CONCLUSION

On a final note, concurrent increases in soil pH and available magnesium were recorded during the flowering and harvest stage of cowpea with a graded dose of magnesium added. A better amelioration of soil pH and the highest available magnesium, potassium, and phosphorus were recorded in soil that received 80 mg.kg⁻¹ of magnesium. But the response of cowpea yield was not following the varying levels of magnesium. A better yield in cowpea was obtained from magnesium at 10 mg.kg⁻¹, suggesting to be the optimum dose for yield maximization.

ACKNOWLEDGEMENTS

The authors hereby acknowledge the financial assistance and research facilities extended by the Kerala Agricultural University.

COMPETING INTERESTS

Authors have declared that no competing interests exist.

REFERENCES

- Chandran P, Ray SK, Bhattacharyya T, Srivastava P, Krishnan P, and Pal DK. Lateritic soils of Kerala, India: their mineralogy, genesis, and taxonomy. Aust. J. Soil Res. 2009;43:839–852.
- 2. Fageria NK. The use of nutrients in crop plants. CRC Press, Boca Raton; 2010.
- Wilkinson SR., Grunes DL, and Sumner ME. Nutrient interactions in soil and plant nutrition. In: Sumner ME, editors. Handbook of soil science. Boca Raton, FL: CRC Press; 2000.
- 4. Gransee A and Führs H. Magnesium mobility in soils as a challenge for soil and plant analysis, magnesium fertilization and root uptake under adverse growth conditions. Plant Soil. 2013; 368:5–21.
- Barber SA. Soil Nutrient Bioavailability: A Mechanistic Approach. John Wiley & Sons. 1995:263.
- Kene HK, Wankhade ST, and Sagare BN. Influence of nutrient spray on yield and oil content of sunflower. Annu. Plant Physiol. 1990:4(2):249-251.
- Camberato JJ. and Pan WL. Bioavailability of calcium, magnesium, and sulfur. In: Sumner, M. E (ed.), Handbook of soil science. Boca Raton, FL: CRC Press; 2000.
- KAU [Kerala Agricultural University]. Package of Practices Recommendations: Crops (15th Ed.). Kerala Agricultural University, Thrissur; 2016.
- 9. Jackson ML. Soil Chemical Analysis. Prentice Hall of India, New Delhi; 1958.
- 10. Walkley AJ. and Black IA. Estimation of soil organic carbon by chromic acid titration method. Soil Sci. 1934;31:29-38.
- 11. Subbiah BV and Asija CL. A rapid procedure for estimation of available nitrogen in soils. Current Sci. 1956;25:259-260.
- Bray RH, and Kurtz LT. Determination of total nitrogen, organic and available form of phosphorus in soils. Soil Sci. 1945;59:39-45.
- Sims JR, and Johnson GV. Micronutrient soil test. In: Mortvedt JJ, Cox FR, Shuman LM, and Welch RM, editors. Micronutrient in Agriculture (2nd Ed.), Madison, U. S. A; 1991.
- 14. Sheoran OP, Tonk DS, Kaushik LS, Hasija RC, and Pannu RS. Statistical Software Package for Agricultural Research Workers. In: Hooda DS, and Hasija RC,

editors. Recent Advances in information theory, Statistics & Computer Applications by Department of Mathematics Statistics, CCS HAU, Hisar; 1998.

- 15. Senbayram M, Gransee A, Wahle V, and Thiel H. Role of magnesium fertilizers in agriculture: plant–soil continuum. Crop Pasture Sci. 2015;66(12):1219-1229.
- 16. Sureshkumar P, Geetha P and Bhindhu PS. Chemistry and Fertility - Management of Humid Tropical Soilsof Kerala as Influenced by Topography and Climate. Indian J. Fertil. 2018;14(3):30-44.
- 17. John I. Wet soil analysis nutrient prescription in paddy soils. M. Sc. (Ag) thesis, Kerala Agricultural University, Thrissur. 2014;130.
- Fageria NK. Optimum soil acidity indices for dry bean production on an Oxisol in notillage system. Commun. Soil Sci. Plant Anal. 2008;39:845-857.
- Adams F. Interactions of phosphorus with other elements in soils and in plants. In: Khasawneh FE, editors, The Role of Phosphorus in Agriculture. Soil

Science Society of America, Madison; 1980.

- 20. Sanyal S, Ghosh D, and Gosh D. Kinetics of potassium adsorption and desorption in some soils of west Bengal. J. Indian Chemical Soc. 2009;86(4):367-376.
- 21. Hannaway DB, Bush LP, and Leggett JE, Mineral composition of Kenhytall fescue as affected by nutrient solution concentrations of Mg and K. J.Plant Nutr. 1982: 137-151.
- 22. Jones JD, Mallarino AP. Influence of source and particle size on agricultural limestone efficiency at increasing soil pH. Soil Sci. Soc. Am. J. 2018;82(1):271-282.
- 23. Berger KC. and Troug E. Boron determination in soil and plant. Indian Eng. Chem. Anal. Ed. 1939;11:540-542.
- 24. Tabatabai MA. Sulfer. In: Page AL, Keeney DR, Baker DE, Miller RH, Roscoe Ellis Jr, and Rhoades JD, editors. Method of Soil Analysis part 2 Chemical and Microbial Properties (2nd Ed.). American Society of Agronomy, Mdison, Wisconsin, USA; 1982.

© 2023 Soniya and Bhindhu; This is an Open Access article distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

> Peer-review history: The peer review history for this paper can be accessed here: https://www.sdiarticle5.com/review-history/102904